# Studies on synthesis of optically active twelve-membered diynes: a convergent construction of a twelve-membered diketodiyne compound with $C_{2}$ symmetry 

Chisato Mukai,* Eiji Kasamatsu, Takao Ohyama and Miyoji Hanaoka*<br>Faculty of Pharmaceutical Sciences, Kanazawa University, Takara-machi, Kanazawa 920-0934, Japan

Received (in Cambridge, UK) 16th November 1999, Accepted 20th December 1999
Published on the Web 7th February 2000


#### Abstract

A convergent procedure for the preparation of optically active twelve-membered diketodiyne derivatives with $C_{2}$ symmetry possessing four hydroxy functionalities was developed starting from the commercially available diethyl L-tartrate. The key step for the construction of the twelve-membered diyne framework involved an intramolecular chromous chloride-mediated coupling reaction between the aldehyde functionality and the alkyne moiety having an iodine atom at its terminus.


## Introduction

Neocarzinostatin (NCS), an antitumor antibiotic, was first isolated from Streptomyces carzinostaticus var F-41 by Ishida et al. in 1965. ${ }^{1}$ In 1985, the chromophoric component of NCS was shown to have a novel bicyclo[7.3.0]dodecadienediyne as a basic skeleton. ${ }^{2}$ The related antitumor antibiotics with a similar strained core framework such as kedarcidin, ${ }^{3} \mathrm{C}-1027,{ }^{4}$ maduropeptin, ${ }^{5}$ and N1999-A2 ${ }^{6}$ have been isolated. Because of the unprecedented unique structure as well as the striking biological activities, much effort has so far been concentrated on the synthesis of the NCS chromophore (1) ${ }^{7}$ and its homologs. ${ }^{8}$ The NCS chromophore (1) can be simplified by ring opening of the epoxy moiety leading to 2 (Scheme 1). The subsequent carbon-carbon bond fission between $\mathrm{C}-1$ and $\mathrm{C}-9$ of 2 would result in the twelve-membered simpler dienediyne derivative 3 with $C_{2}$ symmetry. ${ }^{9}$ We envisioned that this type of twelve-membered compound ${ }^{10}$ would be synthesized by two methods: namely, (i) a coupling reaction of the diyne $\mathbf{4}$ with the diepoxide $\mathbf{5}$, and (ii) by a coupling reaction of $\mathbf{4}$ with the dialdehyde $\mathbf{6}$. Thus, compound $\mathbf{3}$ would be regarded as the significant synthetic intermediate for the construction of the bicyclo[7.3.0]dodecadienediyne skeleton. On the basis of these predictions, the diketodiyne derivatives $\mathbf{7}$ and $\mathbf{8}$, both of which have $C_{2}$ symmetry structure, became our target molecules which would be further manipulated into the corresponding bicyclo[7.3.0]dodecadienediyne via intramolecular aldol type condensation between the $\mathrm{C}-1$ and $\mathrm{C}-9$ positions. In this paper, we describe some results on the synthesis of the optically active twelve-membered diketodiyne derivative with $C_{2}$ symmetry.

## Results and discussion

For the initial evaluation of this strategy, our effort was focused on the development of an efficient method for the construction of $\mathbf{7}$ starting from inexpensive D-mannitol (9) and diethyl L-tartrate (10). At the outset, D-mannitol (9) was converted into the $C_{2}$ symmetric known diepoxide $11^{11}$ (Scheme 2) possessing the vicinal two methoxy functionalities that would be suitable for our purpose. The reactivity of $\mathbf{1 1}$ towards acetylide was preliminarily estimated under the Yamaguchi's conditions. ${ }^{12}$ Thus, treatment of $\mathbf{1 1}$ with the acetylide, prepared from the reaction of phenylacetylene with $n$-butyllithium ( $n$ - BuLi ) in THF in the presence of boron trifluoride-diethyl ether $\left(\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}\right)$ at $-78^{\circ} \mathrm{C}$, unexpectedly afforded the tetrahydro-





3


Scheme 1
furan derivative $\mathbf{1 2}$ in $62 \%$ yield. The formation of $\mathbf{1 2}$ could be interpreted in terms of the ring opening of one of two epoxides by the acetylide in the manner of $\mathrm{S}_{\mathrm{N}} 2$-type displacement ${ }^{13}$ producing the secondary alkoxide species $\mathbf{A}$ which would then intramolecularly attack the remaining epoxide functionality in




12


A

Scheme 2
a 5-exo mode fashion ${ }^{14}$ leading to the formation of $\mathbf{1 2}$. This result strongly indicated that the 3,4 -vicinal hydroxy functionality of the starting material 9 should be protected as some suitable cyclic structure such as an acetonide where two appendages should have a trans relationship. As a result, the unfavorable tetrahydrofuran ring formation observed in the reaction of $\mathbf{1 1}$ with phenylacetylide would be avoided.

The diyne counterpart $\mathbf{1 3}$ for the coupling reaction was prepared from diethyl L-tartrate (10) as follows: the known diethyl $O, O$-dibenzyltartrate (14), derived from $\mathbf{1 0}$ according to the literature, ${ }^{15}$ was reduced with diisobutylaluminium hydride (DIBAL-H) to afford a labile dialdehyde, which was subsequently exposed to Corey's dibromo-olefination conditions ${ }^{16}$ $\left(\mathrm{PPh}_{3}, \mathrm{CBr}_{4}\right)$ to provide $\mathbf{1 5}$ in $61 \%$ yield. The required diyne 13 was then obtained in $68 \%$ yield by exposure of $\mathbf{1 5}$ to $n$ - BuLi in $\mathrm{Et}_{2} \mathrm{O}$. The diepoxide compound 16 possessing the dioxolane skeleton, which is assumed to be a suitable substrate for our purpose on the basis of previous results, was easily prepared from 9 according to the literature ${ }^{17}$ (Scheme 3).


Scheme 3
We attempted the coupling reaction between the diyne compound 13 and the diepoxide derivative 16 (with $C_{2}$ symmetry) under Yamaguchi's conditions in order to construct the twelvemembered ring 17 (Scheme 4). ${ }^{12}$ Treatment of the diyne 13 with $n$ - BuLi in THF at $-78^{\circ} \mathrm{C}$ gave the corresponding lithium acetylide, which was subsequently exposed to $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$. The resulting borane reagent was reacted with the diepoxide 16 at the same temperature to produce the twelve-membered compound 17 in $30 \%$ yield. Although the chemical yield was not satisfactory, we could obtain the twelve-membered compound 17 with $C_{2}$ symmetry. Therefore, our efforts were then directed towards oxidation of $\mathbf{1 7}$ to the corresponding diketo derivative compound 7 depicted in Scheme 1. However, a variety of oxidation conditions afforded only intractable mixtures. No desired compound could be detected in the reaction mixture. At this stage, we tentatively assumed that the dioxolane ring, in other words, the trans-fused five-membered ring would make the
resulting diketone structure rather unstable leading to intractable mixtures. Changing the protecting group at the vicinal $\mathrm{C}-10$ and $\mathrm{C}-11$ positions ${ }^{18}$ from a cyclic species (dioxolane) to an acyclic one, therefore, is necessary to achieve the transformation of $\mathbf{1 7}$ into the diketodiyne compound.

The two hydroxy groups of $\mathbf{1 7}$ were temporarily protected with a benzoyl group under conventional conditions to furnish the dibenzoylated compound $\mathbf{1 8}$ in $66 \%$ yield, which was hydrolyzed with $10 \%$ hydrochloric acid in methanol producing the diol 19 in $70 \%$ yield along with the recovery of the starting material 18 ( $11 \%$ ). Introduction of the (trimethylsilyl)ethoxymethyl (SEM) group, an acyclic group protecting the vicinal $\mathrm{C}-10$ and C-11 hydroxy groups ${ }^{18}$ of $\mathbf{1 9}$, was realized by exposure to SEM chloride in methylene chloride in the presence of diisopropylethylamine (DIPEA) at refluxing temperature to give $\mathbf{2 0}$ in $\mathbf{7 9 \%}$ yield. Upon treatment with potassium carbonate in methanol at room temperature, 20 underwent debenzoylation producing the diol 21 in $87 \%$ yield. The stage was again set for examination of the oxidation of the diol moiety to the corresponding diketo derivative. Thus the diol 21 was exposed to several oxidation conditions as in the case of the oxidation of 17. However, it did not take long to realize that 21 was not a suitable substrate for these oxidation conditions either.

We tentatively interpreted the above results as follows: there are two hydroxy groups in 21, one of which might be oxidized first to afford the keto-alcohol derivative. The remaining hydroxy group of the thus-formed keto-alcohol derivative may attack the carbonyl moiety, before being oxidized, resulting in the formation of the five-membered cyclic acetal species which further collapses under the oxidation conditions to form several undesired compounds. On the basis of these assumptions, we next tried to convert 21 into the mono-protected alcohol derivative. The diol 21 was treated with one equivalent of triethylsilyl (TES) chloride in methylene chloride in the presence of imidazole at room temperature to furnish the desired monoprotected alcohol 22 in $38 \%$ yield together with recovery of the starting material 21 in $52 \%$ yield. We now faced again the examination of the transformation of the hydroxy compound thus prepared into the corresponding keto derivative. After several oxidation conditions were screened, we finally found that oxidation with dimethyl sulfoxide (DMSO) and trifluoroacetic anhydride provided the keto derivative in $81 \%$ yield. The isolated compound from the reaction mixture was, however, found not to be our desired diynone derivative 24, but the allenic keto derivative $\mathbf{2 3},{ }^{19}$ presumably produced through isomerization ${ }^{20}$ from the former during the oxidation reaction. Since it seemed that isomerization of $\mathbf{2 3}$ into the undesired $\mathbf{2 5}$ would be much easier than that into the desired $\mathbf{2 4}$, we turned our endeavor to the preparation of an alternative twelvemembered diketodiyne 8 .

According to our retrosynthetic analysis (Scheme 1), we next investigated a coupling reaction of the diyne 4 with a chiral dialdehyde with $C_{2}$ symmetry (e.g., an antipode of 6). Diethyl L-tartrate (10) was converted into the known dihydroxy compound 26, ${ }^{21}$ which was subsequently treated with tert-butyldimethylsilyl (TBDMS) chloride in THF in the presence of sodium hydride ( NaH ) to afford 27 in $71 \%$ yield (Scheme 5). The coupling reaction of the diyne 13 with the aldehyde freshly derived from 27 was carried out as follows: treatment of $\mathbf{1 3}$ with lithium hexamethyldisilazide (LHMDS) in THF at $-78^{\circ} \mathrm{C}$ in the presence of anhydrous cerium(III) chloride $\left(\mathrm{CeCl}_{3}\right)^{22}$ generated the dicerium acetylide in situ, which was then exposed to the aldehyde, obtained from 27 by Swern oxidation, affording $\mathbf{2 8}$ in $65 \%$ yield as a mixture of two diastereoisomers in a ratio of $59: 41$. The resulting secondary hydroxy group of $\mathbf{2 8}$ was protected with pivaloyl chloride to provide 29 in $83 \%$ yield. Removal of the TBDMS group of $\mathbf{2 9}$ was realized by treatment with acetic acid to afford $\mathbf{3 0}$ in $90 \%$ yield.

After several attempts to construct the twelve-membered framework from 30 via the corresponding aldehyde, we finally


Scheme 5
found that chromous chloride $\left(\mathrm{CrCl}_{2}\right)$-mediated ring closure ${ }^{23}$ was effective. Thus, $\mathbf{3 0}$ was converted into the iodo derivative 31 in $88 \%$ yield by exposure to N -iodosuccinimide (NIS) in the presence of silver nitrate. ${ }^{24}$ The primary alcohol of $\mathbf{3 1}$ was then oxidized with pyridinium chlorochromate (PCC) to provide the corresponding aldehyde 32 in $93 \%$ yield. The stage was now set for the examination of the ring closure mediated by $\mathrm{CrCl}_{2}{ }^{23}$ The aldehyde 32 was reacted with 10 equivalents of $\mathrm{CrCl}_{2}$ in THF in the presence of a catalytic amount of $\mathrm{NiCl}_{2}(5 \mathrm{~mol} \%)$ at room temperature for 24 h to afford the desired twelvemembered compound 33 in $54 \%$ yield. Treatment of 33 with DIBAL-H effected removal of the pivaloyl group to leave the diol $34^{25}$ in $94 \%$ yield. The final phase in this procedure concerned the oxidation of the dihydroxy moieties of the twelve-membered diyne derivative 34. In the cases of the twelve-membered derivatives $\mathbf{1 7}$ and $\mathbf{2 1}$ possessing dihydroxy
functionalities, as previously mentioned (Scheme 4), oxidation under a variety of conditions produced only intractable mixtures. In addition, although we were able to oxidize the hydroxy group of 22, the compound isolated from the reaction mixture was not our desired 24, but 23 in which isomerization of the isolated acetylenic moiety to the conjugated allenic structure had occurred (vide supra). Therefore, it was not clear at this stage if $\mathbf{3 4}$ would be oxidized to give the diketodiyne compound 35. However, PCC oxidation of 34 fortunately proceeded without any difficulty to afford the diketodiyne derivative 35 with $C_{2}$ symmetry in $81 \%$ yield. We are still uncertain about the differences in the reactivity of the oxidizing agents with 34 and the other twelve-membered dihydroxy derivatives $\mathbf{1 7}$ and $\mathbf{2 1}$.

In summary, we have developed a procedure for the preparation of the optically active twelve-membered diketodiyne derivative $\mathbf{3 5}$ with $C_{2}$ symmetry starting from the commercially
available diethyl L-tartrate (10). Since the diketodiyne derivative 35 possesses two benzoyloxy groups at the C-4 and C-5 positions from which the epoxy functionality may be formed, it is regarded as a suitable precursor for the bicyclo[7.3.0]dodecadienediyne framework, a core of the NCS chromophore 1. Attempts to improve the chemical yields of $\mathbf{3 5}$ from $\mathbf{1 0}$ and to convert 35 into the corresponding bicyclo[7.3.0]dodecadienediyne through aldol-type condensation ${ }^{26}$ are now in progress.

## Experimental

IR spectra were measured with a JASCO A-102 spectrometer in $\mathrm{CHCl}_{3}$ and mass spectra with Hitachi M-80 and JEOL JMS-SX 102 A mass spectrometers. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on JEOL EX-270 and JEOL JNM-GX 500 spectrometers, using $\mathrm{CDCl}_{3}$ as solvent and either tetramethylsilane as internal standard for compounds that have no silyl group, or $\mathrm{CHCl}_{3}$ ( 7.26 ppm ) for compounds possessing the silyl group. ${ }^{13} \mathrm{C}$ NMR spectra were recorded on JEOL EX-270 and JEOL JNM-GX 500 spectrometers in $\mathrm{CDCl}_{3}$ with $\mathrm{CDCl}_{3}$ ( 77.0 ppm ) as an internal reference. All $J$ values are in Hz and $[a]_{\mathrm{D}}$ values in $10^{-1}$ deg $\mathrm{cm}^{2} \mathrm{~g}^{-1}$. $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was freshly distilled from $\mathrm{P}_{2} \mathrm{O}_{5}$, and THF, toluene and $\mathrm{Et}_{2} \mathrm{O}$ from sodium-benzophenone prior to use. All reactions were carried out under a nitrogen atmosphere. Silica gel (Silica gel 60, 230-400 mesh, Merck) was used for chromatography. Organic extracts were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$.

## ( $2 R, 3 S, 4 S, 5 S$ )-5-Hydroxymethyl-3,4-dimethoxy-2-(3-phenyl-prop-2-yn-1-yl)tetrahydrofuran (-)-12

A solution of $n$-BuLi in hexane $\left(1.41 \mathrm{~mol} \mathrm{dm}^{-3} ; 0.58 \mathrm{~cm}^{3}, 0.82\right.$ mmol ) was added to a solution of phenylacetylene ( 84.0 mg , $0.82 \mathrm{mmol})$ in THF $\left(5.5 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$. After being stirred for $10 \mathrm{~min}, \mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}\left(0.10 \mathrm{~cm}^{3}, 0.82 \mathrm{mmol}\right)$ was added to the reaction mixture and stirring was continued for 10 min at the same temperature. A solution of $\mathbf{1 1}(47.8 \mathrm{mg}, 0.27 \mathrm{mmol})$ in THF $\left(1.5 \mathrm{~cm}^{3}\right)$ was then added to the reaction mixture, which was further stirred for 1 h , quenched by addition of saturated aq. $\mathrm{NaHCO}_{3}$ and extracted with AcOEt. The extract was washed with water and brine, dried and concentrated to dryness. The residue was chromatographed with hexane-AcOEt (3:1) to give ( - ) - $\mathbf{1 2}\left(47.2 \mathrm{mg}, 62 \%\right.$ ) as a colorless oil (Found: $\mathrm{M}^{+}$, 276.1361. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{4}$ requires $M$, 276.1362); $[\alpha]_{\mathrm{D}}^{26}-2.6(c 0.50$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 3524(\mathrm{OH})$ and $2225(\mathrm{C} \equiv \mathrm{C}) ; \delta_{\mathrm{H}} 7.42-$ $7.40(2 \mathrm{H}, \mathrm{m}$, aromatic H$), 7.30-7.26(3 \mathrm{H}, \mathrm{m}$, aromatic H$)$, 4.13-4.03 $(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 3.89-3.83(4 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$, $4-\mathrm{H}$ and $\left.\mathrm{C}-5-\mathrm{CH}_{2}\right), 3.46(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.44(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $2.83(1 \mathrm{H}, \mathrm{dd}, J 17.1$ and $4.9, \mathrm{C}-2-\mathrm{CH}), 2.75(1 \mathrm{H}, \mathrm{dd}, J 17.1$ and $8.3, \mathrm{C}-2-\mathrm{CH})$ and $2.34(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}) ; \delta_{\mathrm{C}} 131.6,128.2$, $127.9,123.4,86.6,86.4,85.8,82.3,81.0,80.5,61.6,57.7,57.4$ and 24.5; m/z $276\left(\mathrm{M}^{+}, 24 \%\right), 244$ (69), 213 (60), 115 (65) and 99 (100).

## (3S,4S)-3,4-Bis(benzyloxy)-1,1,6,6-tetrabromohexa-1,5-diene (+)-15

To a solution of $\mathbf{1 4}(7.00 \mathrm{~g}, 18.1 \mathrm{mmol})$ in toluene $\left(36 \mathrm{~cm}^{3}\right)$ was added a solution of DIBAL-H in hexane ( $40.4 \mathrm{~cm}^{3}, 0.95 \mathrm{~mol}$ $\mathrm{dm}^{-3} ; 38.0 \mathrm{mmol}$ ) at $-78^{\circ} \mathrm{C}$. The reaction mixture was kept for 1 h at the same temperature. To a solution of $\mathrm{PPh}_{3}(28.5 \mathrm{~g}, 108$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(50 \mathrm{~cm}^{3}\right)$ was added a solution of $\mathrm{CBr}_{4}(18.0 \mathrm{~g}$, $54.3 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(10 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$ and the solution was stirred for an additional 10 min . A solution of the crude dialdehyde derivative, freshly prepared from 14 in toluene, was then added to a solution of the ylide in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution at $-78^{\circ} \mathrm{C}$. The reaction mixture was then gradually warmed to rt and stirring was continued for 16 h at rt , quenched by addition of saturated aq. $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The filtrate was passed through a short pad of silica gel with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in order to remove
triphenylphosphine oxide. The residue was chromatographed with hexane-AcOEt $(50: 1)$ to give $(+)-15(6.69 \mathrm{~g}, 61 \%)$ as a pale yellow oil (Found: C, 39.2; H, 3.0. $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{Br}_{4} \mathrm{O}_{2}$ requires C, $39.4 ; \mathrm{H}, 3.0 \%) ;[a]_{\mathrm{D}}^{25}+45.3\left(c 1.0, \mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1} 1690$ $(\mathrm{C}=\mathrm{C})$ and $1610(\mathrm{C}=\mathrm{C}) ; \delta_{\mathrm{H}} 7.37-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 6.60$ ( $2 \mathrm{H}, \mathrm{d}, J 7.32,2-\mathrm{H}$ and $5-\mathrm{H}$ ), 4.68, $4.43(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 12.2$, benzylic H) and $4.15(2 \mathrm{H}, \mathrm{d}, J 7.3,3-\mathrm{H}$ and $4-\mathrm{H})$; $\delta_{\mathrm{C}} 137.3$, 136.0, 128.4, 128.0, 127.9, 92.8, 79.9 and $71.6 ; \mathrm{m} / \mathrm{z} 606\left(\mathrm{M}^{+}\right.$, $0.9 \%$ ), 607 (4.1), 529 (15), 527 (4.9), 305 (48), 195 (36), 115 (66) and 91 (100).

## (3S,4S)-3,4-Bis(benzyloxy)hexa-1,5-diyne (+)-13

To a solution of $\mathbf{1 5}(3.80 \mathrm{~g}, 6.24 \mathrm{mmol})$ in dry $\mathrm{Et}_{2} \mathrm{O}\left(21 \mathrm{~cm}^{3}\right)$ was added $n$-BuLi in hexane ( $1.63 \mathrm{~mol} \mathrm{dm}^{-3} ; 16.1 \mathrm{~cm}^{3}, 26.2 \mathrm{mmol}$ ) at $-78^{\circ} \mathrm{C}$ and the reaction mixture was stirred for 30 min , quenched by addition of saturated aq. $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted with AcOEt. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (50:1) afforded (+)-13 (1.23 g, $68 \%$ ) as a colorless oil (Found: C, 82.8; H, 6.3. $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{2}$ requires C, 82.7; H, $6.25 \%$ ); $[a]_{\mathrm{D}}^{27}+134.8\left(c 0.51, \mathrm{CHCl}_{3}\right) ; v_{\max } /$ $\mathrm{cm}^{-1} 3300(\mathrm{C} \equiv \mathrm{C}-\mathrm{H})$ and $2100(\mathrm{C} \equiv \mathrm{C}) ; \delta_{\mathrm{H}} 7.35-7.32(10 \mathrm{H}, \mathrm{m}$, aromatic H), $4.85,4.62(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 11.7$, benzylic H), 4.31 $(2 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ and $4-\mathrm{H})$ and $2.57(2 \mathrm{H}, \mathrm{s}, 1-\mathrm{H}$ and $6-\mathrm{H}) ; \delta_{\mathrm{C}} 137.1$, $128.3,128.0,127.8,79.2,75.8,70.9$ and $70.5 ; \mathrm{m} / \mathrm{z} 290\left(\mathrm{M}^{+}\right.$, $0.7 \%), 199(86), 107$ (75), 91 (100) and 65 (81).

## ( $1 R, 2 R, 3 R, 4 R, 8 S, 9 S$ )-8,9-Bis(benzyloxy)-2,3-(isopropylidene-dioxy)cyclododeca-6,10-diyne-1,4-diol (+)-17

According to the procedure described for the preparation of 12, the diyne derivative $13(823 \mathrm{mg}, 2.84 \mathrm{mmol})$ in THF $\left(70 \mathrm{~cm}^{3}\right)$ was successively treated with a solution of $n$ - BuLi in hexane $\left(1.46 \mathrm{~mol} \mathrm{dm}^{-3} ; 5.83 \mathrm{~cm}^{3}, 8.51 \mathrm{mmol}\right)$ and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}\left(1.07 \mathrm{~cm}^{3}\right.$, $8.51 \mathrm{mmol})$ at $-78^{\circ} \mathrm{C}$. A solution of $\mathbf{1 6}(528 \mathrm{mg}, 2.84 \mathrm{mmol})$ in THF ( $10.0 \mathrm{~cm}^{3}$ ) was added to the borane reagent in THF and the mixture was stirred for 1 h , quenched by addition of saturated aq. $\mathrm{NaHCO}_{3}$ and extracted with AcOEt. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt gave $(+)-17(408 \mathrm{mg}, 30 \%)$ as a colorless oil (Found: $\mathrm{M}^{+}, 476.2192$. $\mathrm{C}_{29} \mathrm{H}_{32} \mathrm{O}_{6}$ requires $M, 476.2199$ ); $[a]_{\mathrm{D}}^{26}+84.3$ ( c 0.51, $\mathrm{CHCl}_{3}$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 3525(\mathrm{OH})$ and $2220(\mathrm{C} \equiv \mathrm{C}) ; \delta_{\mathrm{H}} 7.37-7.23(10 \mathrm{H}, \mathrm{m}$, aromatic H), 4.78, $4.59(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 11.7$, benzylic H), 4.46 ( $2 \mathrm{H}, \mathrm{d}, J 2.9,2-\mathrm{H}$ and $3-\mathrm{H}$ ), $4.37(2 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ and $9-\mathrm{H}), 3.93$ $(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $4-\mathrm{H}), 2.78(2 \mathrm{H}, \mathrm{dd}, J 17.1$ and $6.8,5-\mathrm{H}$ and $12-\mathrm{H}), 2.64(2 \mathrm{H}, \mathrm{d}, J 17.1,5-\mathrm{H}$ and $12-\mathrm{H}), 2.44(2 \mathrm{H}, \mathrm{br}$ s, OH$)$ and $1.41(6 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}} 137.4,128.3,127.9,127.7,108.9,84.6$, 80.6, 78.4, 72.7, 71.2, 70.0, 27.0, 23.5; m/z 476 ( $\left.\mathrm{M}^{+}, 2.3 \%\right), 461$ (21), 385 (17), 91 (100), 65 (26) and 59 (31).

## ( $1 R, 2 S, 3 S, 4 R, 8 S, 9 S$ )-1,4-Bis(benzoyloxy)-8,9-bis(benzyloxy)-2,3-(isopropylidenedioxy)cyclododeca-6,10-diyne (+)-18

To a solution of $17(66.0 \mathrm{mg}, 0.14 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(87.0 \mathrm{mg}$, $0.83 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(2.0 \mathrm{~cm}^{3}\right)$ was added a solution of benzoyl chloride ( $85.0 \mathrm{mg}, 0.55 \mathrm{mmol}$ ) at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 18 h at rt , quenched by addition of water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed with water and brine, dried and concentrated to dryness. The residue was chromatographed with hexane-AcOEt (8:1) to give (+)-18 $(61.5 \mathrm{mg}, 66 \%)$ as a colorless solid, $\mathrm{mp} 54-56^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 75.3; H, 5.9. $\mathrm{C}_{43} \mathrm{H}_{40} \mathrm{O}_{8}$ requires C, $75.4 ; \mathrm{H}, 5.9 \%$ ); $[a]_{\mathrm{D}}^{26}+29.0\left(c 0.26, \mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1} 2250(\mathrm{C} \equiv \mathrm{C}), 1719(\mathrm{CO}) ; \delta_{\mathrm{H}}$ $8.10(4 \mathrm{H}, \mathrm{d}, J 7.3$, aromatic H), $7.54(2 \mathrm{H}, \mathrm{t}, J 7.3$, aromatic H), $7.42-7.26(14 \mathrm{H}, \mathrm{m}$, aromatic H$), 5.35(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $4-\mathrm{H})$, $4.90(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $3-\mathrm{H}), 4.75,4.56(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 11.7$, benzylic H), $4.38(2 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ and $9-\mathrm{H}), 3.09(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ and $12-\mathrm{H}), 2.71(2 \mathrm{H}, \mathrm{d}, J 17.1,5-\mathrm{H}$ and $12-\mathrm{H})$ and $1.47(6 \mathrm{H}, \mathrm{s}, \mathrm{Me})$; $\delta_{\mathrm{C}} 165.3,137.3,133.2,129.8,129.5,128.4,128.2,127.8,127.6$,
110.2, 83.6, 79.8, 72.4, 72.0, 70.9, 70.9, 27.2 and 20.9; m/z 684 $\left(\mathrm{M}^{+}, 0.6 \%\right), 669$ (4.7), 535 (5.9), 507 (5.1), 456 (5.5), 291 (5.2), 105 (100), 91 (95) and 77 (22).

## (1R,2S,3S,4R,8S,9S)-1,4-Bis(benzoyloxy)-8,9-bis(benzyloxy)-cyclododeca-6,10-diyne-2,3-diol (+)-19

A solution of $\mathbf{1 8}(83.0 \mathrm{mg}, 0.12 \mathrm{mmol})$ in $\mathrm{MeOH}\left(2.0 \mathrm{~cm}^{3}\right)$ and THF ( $2.00 \mathrm{~cm}^{3}$ ) in the presence of conc. $\mathrm{HCl}\left(1.0 \mathrm{~cm}^{3}\right)$ was stirred for 24 h at rt and the solvent was evaporated off. The residue was taken up in AcOEt, which was washed with $10 \%$ aq. NaOH , water and brine, dried and concentrated to dryness. The residue was chromatographed with hexane-AcOEt (2:1) to give (+)-19 ( $55.0 \mathrm{mg}, 70 \%$ ) as a colorless solid, $\mathrm{mp} 46-47^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 74.2; H, 5.6 . $\mathrm{C}_{40} \mathrm{H}_{36} \mathrm{O}_{8}$ requires $\mathrm{C}, 74.5 ; \mathrm{H}, 5.6 \%$ ); $[a]_{\mathrm{D}}^{25}+12.3$ (c 0.50 , $\left.\mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1} 3525(\mathrm{OH}), 2220(\mathrm{C} \equiv \mathrm{C})$ and $1719(\mathrm{CO})$; $\delta_{\mathrm{H}} 8.07-8.05(4 \mathrm{H}, \mathrm{m}$, aromatic H$), 7.54(2 \mathrm{H}, \mathrm{m}$, aromatic H$)$, 7.39-7.27 $(14 \mathrm{H}, \mathrm{m}$, aromatic H), $5.25(2 \mathrm{H}, \mathrm{m}, 1 \mathrm{H}$ and $4-\mathrm{H})$, $4.80(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $3-\mathrm{H}), 4.77,4.56(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, ~ J 11.7$, benzylic H), $4.36(2 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ and $9-\mathrm{H}), 3.16(2 \mathrm{H}, \mathrm{d}, J 6.8$, $\mathrm{OH}), 3.02(2 \mathrm{H}, \mathrm{dd}, J 17.6$ and $5.9,5-\mathrm{H}$ and $12-\mathrm{H})$ and 2.82 $(2 H, m, 5-H$ and $12-H) ; \delta_{\mathrm{C}} 166.0,137.3,133.4,129.7,129.6$, $128.6,128.4,127.9,127.8,83.4,80.1,73.7,72.2,71.0,69.2$ and 21.8; m/z $644\left(\mathrm{M}^{+}, 0.8 \%\right), 614(0.8), 535(7.4), 307(14), 105$ (100), 91 (100) and 77 (52).
(1R,2S,3S,4R,8S,9S)-1,4-Bis(benzoyloxy)-2,3-bis[(trimethyl-silylethoxymethyl)oxy]-8,9-bis(benzyloxy)cyclododeca-6,10diyne (+)-20
SEMCl ( $179 \mathrm{mg}, 1.06 \mathrm{mmol}$ ) was added to a solution of $\mathbf{1 9}$ $(68.0 \mathrm{mg}, 0.11 \mathrm{mmol})$ and DIPEA ( $196 \mathrm{mg}, 2.12 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1.0 \mathrm{~cm}^{3}\right)$ at rt . After being refluxed for 5 h , the reaction mixture was quenched by addition of water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (13:1) afforded $(+)-\mathbf{2 0}(75.0 \mathrm{mg}, 79 \%)$ as a colorless oil (Found: C, 68.8; H, 7.3. $\mathrm{C}_{52} \mathrm{H}_{64} \mathrm{O}_{10} \mathrm{Si}_{2}$ requires $\mathrm{C}, 69.0 ; \mathrm{H}, 7.1 \%$ ); $[a]_{\mathrm{D}}^{27}+60.8(c 0.50$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 2230(\mathrm{C} \equiv \mathrm{C})$ and $1719(\mathrm{CO}) ; \delta_{\mathrm{H}} 8.06-7.26$ $(20 \mathrm{H}, \mathrm{m}$, aromatic H$), 5.45(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $4-\mathrm{H}), 4.87$, $4.85\left(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 6.8, \mathrm{OCH}_{2} \mathrm{O}\right), 4.80,4.59(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 11.7$, benzylic H ), $4.67(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $3-\mathrm{H}), 4.41(2 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ and $9-\mathrm{H}), 3.71-3.55\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH}_{2} \mathrm{CH}_{2}\right), 2.99(2 \mathrm{H}, \mathrm{dd}, J 17.1$ and $6.8,5-\mathrm{H}$ and $12-\mathrm{H}), 2.86(2 \mathrm{H}, \mathrm{d}, J 17.1,5-\mathrm{H}$ and $12-\mathrm{H})$, $0.89-0.85\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH} \mathrm{CH}_{2}\right)$ and $-0.05(18 \mathrm{H}, \mathrm{s}, \mathrm{Me})$; $\delta_{\mathrm{C}} 165.5,137.5,133.1,129.9,129.8,128.4,128.3,127.9,127.7$, $96.4,83.9,79.9,76.0,72.8,72.5,70.9,66.2,21.2,17.9$ and $-1.5 ;$ FABMS $m / z 927\left(\mathrm{M}^{+}+23,0.6 \%\right), 105$ (50), 91 (52), 73 (100) and 55 (17).

## ( $1 R, 2 R, 3 R, 4 R, 8 S, 9 S)$-2,3-Bis[(trimethylsilylethoxymethyl)-oxy]-8,9-bis(benzyloxy)cyclododeca-6,10-diyne-1,4-diol (+)-21

$\mathrm{K}_{2} \mathrm{CO}_{3}(34.0 \mathrm{mg}, 0.25 \mathrm{mmol})$ was added to a solution of $\mathbf{2 0}$ $(56.0 \mathrm{mg}, 0.06 \mathrm{mmol})$ in $\mathrm{MeOH}\left(4.0 \mathrm{~cm}^{3}\right)$ and the mixture was stirred at rt for 12 h . MeOH was evaporated off and the residue was taken up in AcOEt, which was washed with saturated aq. $\mathrm{NH}_{4} \mathrm{Cl}$, water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (4:1) afforded (+)-21 ( $38.0 \mathrm{mg}, 87 \%$ ) as a colorless oil (Found: $\mathrm{C}, 65.2 ; \mathrm{H}, 8.2 . \mathrm{C}_{38} \mathrm{H}_{56} \mathrm{O}_{8} \mathrm{Si}_{2}$ requires $\mathrm{C}, 65.5 ; \mathrm{H}$, $8.1 \%) ;[a]_{\mathrm{D}}^{27}+10.2\left(c 0.51, \mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 3523(\mathrm{OH})$ and $2220(\mathrm{C} \equiv \mathrm{C}) ; \delta_{\mathrm{H}} 7.37-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 4.94,4.79$ $\left(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 6.8, \mathrm{OCH}_{2} \mathrm{O}\right), 4.76,4.59(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 12.2$, benzylic H), $4.43(2 \mathrm{H}, \mathrm{d}, J 3.9,2-\mathrm{H}$ and $3-\mathrm{H}), 4.36(2 \mathrm{H}, \mathrm{s}$, $8-\mathrm{H}$ and $9-\mathrm{H}), 4.26(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $4-\mathrm{H}), 3.75-3.69(4 \mathrm{H}, \mathrm{m}$, $\mathrm{TMSCH}_{2} \mathrm{CH}_{2}$ ), $3.52(2 \mathrm{H}, \mathrm{d}, J 7.8, \mathrm{OH}), 2.80(2 \mathrm{H}, \mathrm{dd}, J 16.6$ and $5.4,5-\mathrm{H}$ and $12-\mathrm{H}), 2.56(2 \mathrm{H}, \mathrm{dd}, J 16.6$ and $9.8,5-\mathrm{H}$
and $12-\mathrm{H}), 0.98-0.94\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH} \mathrm{CH}_{2}\right)$ and $0.02(18 \mathrm{H}$, s , Me); $\delta_{\mathrm{C}} 137.5,128.3,127.8,127.7,93.9,83.3,79.7,72.9$, 71.3, 69.1, 66.8, 24.2, 18.1 and $-1.4 ; \mathrm{m} / \mathrm{z} 696\left(\mathrm{M}^{+}, 1.2 \%\right)$, 548 (6.0), 429 (12), 91 (100) and 73 (100).
( $1 R, 2 R, 3 S, 4 R, 8 S, 9 S)$-4-Triethylsiloxy-2,3-bis[(trimethylsilyl-ethoxymethyl)oxy]-8,9-bis(benzyloxy)cyclododeca-6,10-diyn-1ol (+)-22

TESCl ( $6.20 \mathrm{mg}, 0.04 \mathrm{mmol}$ ) was added to a solution of $\mathbf{2 1}$ ( $21.0 \mathrm{mg}, 0.03 \mathrm{mmol}$ ) and imidazole ( $10.0 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(0.5 \mathrm{~cm}^{3}\right)$ and the mixture was stirred at rt for 7 h , quenched by addition of water and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt afforded ( + )-22 ( $9.30 \mathrm{mg}, 38 \%$ ) and recovery of the starting material 21 ( $11.0 \mathrm{mg}, 52 \%$ ). Compound ( + )-22 was obtained as a colorless oil (Found: $\mathrm{M}^{+}+23$, 833.4279. $\mathrm{C}_{44} \mathrm{H}_{70} \mathrm{O}_{8} \mathrm{Si}_{3} \mathrm{Na}$ requires $M+23$, 833.4276); $[a]_{\mathrm{D}}^{26}+82.7$ (c 0.20 , $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 3520(\mathrm{OH})$ and $2220(\mathrm{C}=\mathrm{C}) ; \delta_{\mathrm{H}} 7.36-7.25$ $(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 4.89-4.56(8 \mathrm{H}, \mathrm{m}$, benzylic H and $\left.\mathrm{OCH}_{2} \mathrm{O}\right), 4.45(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 4.34(2 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ and $9-\mathrm{H}), 4.16$ $(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 4.01(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 3.96(1 \mathrm{H}, \mathrm{dd}, J 7.3$ and 4.4 , $3-\mathrm{H}), 3.80-3.54\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH}_{2} \mathrm{CH}_{2}\right), 3.46(1 \mathrm{H}, \mathrm{m}, \mathrm{OH})$, $2.76-2.54(4 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ and $12-\mathrm{H}), 0.98(9 \mathrm{H}, \mathrm{t}, J 7.8, \mathrm{Me}), 0.98-$ $0.93\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH}_{2} \mathrm{CH}_{2}\right), 0.64\left(6 \mathrm{H}, \mathrm{q}, J 7.8, \mathrm{MeCH}_{2}\right), 0.02$ $(9 \mathrm{H}, \mathrm{s}, \mathrm{Me})$ and $0.01(9 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}} 137.7,137.6,128.3,128.0$, 127.9, 127.6, 96.8, 96.7, 85.3, 85.1, 79.7, 79.6, 76.8, 76.3, 72.7, $72.5,72.2,70.9,70.9,70.6,66.3,66.1,24.9,23.4,18.2,18.1,6.8$, $4.9,-1.4$ and -1.5 ; FABMS $m / z 833\left(\mathrm{M}^{+}+23,0.3 \%\right), 226$ (4.8), 91 (48) and 73 (100).

## ( $2 S, 3 S, 4 R, 8 S, 9 S$ )-4-Triethylsiloxy-2,3-bis[(trimethylsilyl-ethoxymethyl)oxy]-8,9-bis(benzyloxy)cyclododeca-10,11-dien-6-yn-1-one (+)-23

A solution of DMSO ( $76.0 \mathrm{mg}, 1.02 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(0.5 \mathrm{~cm}^{3}\right)$ was gradually added to a solution of $\left(\mathrm{CF}_{3} \mathrm{CO}\right)_{2} \mathrm{O}(104 \mathrm{mg}, 0.51$ mmol) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(0.5 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$. After stirring of the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution for 15 min , a solution of the alcohol $22(81.0$ $\mathrm{mg}, 0.10 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(0.5 \mathrm{~cm}^{3}\right)$ was added and the reaction mixture was stirred at the same temperature for $1 \mathrm{~h} . \mathrm{Et}_{3} \mathrm{~N}(152$ $\mathrm{mg}, 1.53 \mathrm{mmol}$ ) was added to the reaction mixture, which was then gradually warmed to rt and quenched by addition of water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane- $\operatorname{AcOEt}(20: 1)$ afforded ( + )-23 (65.0 $\mathrm{mg}, 81 \%$ ) as a colorless oil (Found: $\mathrm{M}^{+}+23,831.4113$. $\mathrm{C}_{44} \mathrm{H}_{68} \mathrm{O}_{8} \mathrm{Si}_{3} \mathrm{Na}$ requires $\left.M+23,831.4120\right) ;[a]_{\mathrm{D}}^{23}+73.4(c 0.20$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 2230(\mathrm{C} \equiv \mathrm{C}), 1952(=\mathrm{C}=)$ and $1675(\mathrm{CO})$; $\delta_{\mathrm{H}} 7.37-7.27(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 5.97(1 \mathrm{H}, \mathrm{dd}, J 6.3$ and 3.3 , $12-\mathrm{H}), 5.91(1 \mathrm{H}, \mathrm{dd}, J 6.3$ and $4.6,10-\mathrm{H}), 5.11(1 \mathrm{H}, \mathrm{d}, J 7.9$, $2-\mathrm{H}), 4.87-4.48\left(9 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}\right.$, benzylic H and $\left.\mathrm{OCH}_{2} \mathrm{O}\right), 4.38$ $(1 \mathrm{H}, \mathrm{d}, J 5.9,8-\mathrm{H}), 4.32(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 4.23(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 3.73-$ $3.46\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH}_{2} \mathrm{CH}_{2}\right), 2.90(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.38(1 \mathrm{H}, \mathrm{m}$, $5-\mathrm{H}), 0.96(9 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{Me}), 0.96-0.79\left(4 \mathrm{H}, \mathrm{m}, \mathrm{TMSCH} \mathrm{CH}_{2}\right)$, $0.60\left(6 \mathrm{H}, \mathrm{q}, J 7.6, \mathrm{CH}_{2} \mathrm{Me}\right), 0.00(9 \mathrm{H}, \mathrm{s}, \mathrm{Me})$ and $-0.02(9 \mathrm{H}, \mathrm{s}$, Me); $\delta_{\mathrm{C}} 213.6,197.6,137.6,175.5,128.4,128.3,127.9,127.8$, 127.8, 127.7, 99.0, 96.5, 95.8, 95.3, 85.6, 81.3, 78.7, 76.3, 75.8, $72.0,71.4,70.7,70.6,65.6,65.4,22.8,18.0,17.9,6.8,4.7,-1.4$ and -1.5 ; FABMS $m / z 831\left(\mathrm{M}^{+}+23,2.4 \%\right), 226$ (6.8), 91 (84), 87 (24), 73 (100) and 59 (17).

## (3S,4S)-6-(tert-Butyldimethylsiloxy)-3,4-bis(methoxy)hexan-1-

 ol ( - )- 27A solution of $26(1.99 \mathrm{~g}, 11.4 \mathrm{mmol})$ in THF ( $10.0 \mathrm{~cm}^{3}$ ) was added to a suspension of $\mathrm{NaH}(303 \mathrm{mg}, 12.6 \mathrm{mmol})$ in THF $\left(30.0 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$. The THF solution was stirred for 1 h , to which a solution of TBDMSCl $(1.90 \mathrm{~g}, 12.6 \mathrm{mmol})$ in THF $(10.0 \mathrm{ml})$ was added. After stirring for 3 h , the reaction mixture
was quenched by addition of water and extracted with AcOEt. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (5:1) afforded ( - )-27 ( $2.34 \mathrm{~g}, 71 \%$ ) as a colorless oil (Found: C, 57.7; H, 10.9. $\mathrm{C}_{14} \mathrm{H}_{32} \mathrm{O}_{4}$ Si requires C, 57.5 ; $\mathrm{H}, 11.0 \%)$; $[a]_{\mathrm{D}}^{26}-46.6\left(c 0.50, \mathrm{CHCl}_{3}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 3450(\mathrm{OH})$; $\delta_{\mathrm{H}} 3.76-3.43(6 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}$ and $6-\mathrm{H}), 3.42(6 \mathrm{H}, \mathrm{s}, \mathrm{Me})$, $2.70(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.82-1.53(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 0.89$ $\left(9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right), 0.05(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$ and $0.04(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}} 81.0,77.5$, $61.0,59.4,58.5,58.0,32.7,32.1,25.9,18.2,-5.4$ and -5.4 ; CIMS $m / z 293\left(\mathrm{M}^{+}+1,100\right), 261$ (18), 203 (25), 161 (24) and 89 (84).
$(3 S, 4 S, 6 R, 9 S, 10 S)$ - and ( $3 S, 4 S, 6 S, 9 S, 10 S$ )-9,10-Bis(benzyl-oxy)-1-(tert-butyldimethysiloxy)-3,4-dimethoxydodeca-7,11-diyn-6-ol 28
A solution of DMSO ( $470 \mathrm{mg}, 5.97 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1.0 \mathrm{~cm}^{3}\right)$ was gradually added to a solution of oxalyl chloride ( 378 mg , 2.99 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(8.0 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$. After stirring of the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution for 15 min , a solution of the alcohol 27 (291 $\mathrm{mg}, 1.00 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(2.0 \mathrm{~cm}^{3}\right)$ was added and the reaction was stirred at $-78^{\circ} \mathrm{C}$ for $1 \mathrm{~h} . \mathrm{Et}_{3} \mathrm{~N}(91.0 \mathrm{mg}, 8.96 \mathrm{mmol})$ was added to the reaction mixture, which was then gradually warmed to rt and quenched by addition of water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed with water and brine, dried and concentrated to leave the crude aldehyde. A solution of diyne $13(866 \mathrm{mg}, 2.99 \mathrm{mmol})$ in THF $\left(2.0 \mathrm{~cm}^{3}\right)$ was added to a suspension of anhydrous $\mathrm{CeCl}_{3}(1.96 \mathrm{~g}, 7.97 \mathrm{mmol})$ in THF $\left(25.0 \mathrm{~cm}^{3}\right)$. The mixture was stirred for 5 min , then LHMDS ( $0.95 \mathrm{~mol} \mathrm{dm}^{-3} ; 6.29 \mathrm{~cm}^{3}, 5.97 \mathrm{mmol}$ ) was added to the mixture, and the stirring was continued for 1 h at $-78^{\circ} \mathrm{C}$. A solution of the crude aldehyde, prepared from 27, in THF ( $3.0 \mathrm{~cm}^{3}$ ) was then added to the solution of cerium acetylide in THF. The reaction mixture was stirred for 1 h , quenched by addition of saturated aq. $\mathrm{NH}_{4} \mathrm{Cl}$ and filtered through Celite. The filtrate was diluted with AcOEt, which was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (4:1) afforded $28(380 \mathrm{mg}$, $65 \%$ ) as a colorless oil (Found: C, 70.0; H, 8.5. $\mathrm{C}_{34} \mathrm{H}_{48} \mathrm{O}_{6} \mathrm{Si}$ requires C, $70.3 ; \mathrm{H}, 8.3 \%)$; $v_{\text {max }} / \mathrm{cm}^{-1} 3460(\mathrm{OH}), 3306(\mathrm{C}=\mathrm{C}-\mathrm{H})$ and $2120(\mathrm{C} \equiv \mathrm{C}) ; \delta_{\mathrm{H}} 7.39-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$)$, 4.84-4.81 $(2 \mathrm{H}, \mathrm{m}$, benzylic H$), 4.69-4.66(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 4.64-4.59(2 \mathrm{H}$, m , benzylic H), $4.38(1 \mathrm{H}$, dd, $J 6.3$ and $1.5,9-\mathrm{H})$, $4.31(1 \mathrm{H}$, dd, $J 6.3$ and $2.4,10-\mathrm{H}), 3.96-3.54(4 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}, 3-\mathrm{H}$ and $4-\mathrm{H})$, $3.45\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.39\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.39\left(\frac{41}{100} \times 3 \mathrm{H}\right.$, $\mathrm{s}, \mathrm{Me}), 3.38\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 2.55\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{d}, J 2.4,12-\mathrm{H}\right)$, $2.54\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{d}, J 2.4,12-\mathrm{H}\right), 2.01-1.45(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 0.90\left(9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right), 0.06(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$ and $0.06(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$; $m / z 580\left(\mathrm{M}^{+}, 0.2\right), 491$ (8.5), 145 (48), 91 (89), 89 (100) and 73 (32).

## ( $3 S, 4 S, 6 R, 9 S, 10 S$ )- and ( $3 S, 4 S, 6 S, 9 S, 10 S$ )-9,10-Bis(benzyl-oxy)-1-(tert-butyldimethylsiloxy)-3,4-dimethoxy-6-(pivaloyloxy)-dodeca-7,11-diyne 29

$\operatorname{PivCl}(137 \mathrm{mg}, 1.12 \mathrm{mmol})$ was added to a solution of $\mathbf{2 8}$ ( 325 $\mathrm{mg}, 0.56 \mathrm{mmol}), \mathrm{Et}_{3} \mathrm{~N}(166 \mathrm{mg}, 1.68 \mathrm{mmol})$ and DMAP ( 10.0 $\mathrm{mg}, 0.08 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(6.0 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$. After being stirred for 4.5 h at rt , the reaction mixture was quenched by addition of water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was washed with water and brine, dried and concentrated to dryness. Chromatography of the residue with hexane-AcOEt (13:1) afforded $29(308 \mathrm{mg}, 83 \%)$ as a colorless oil (Found: $\mathrm{M}^{+}, 664.3796$. $\mathrm{C}_{39} \mathrm{H}_{56} \mathrm{O}_{7} \mathrm{Si}$ requires $\left.M, 664.3795\right)$; $v_{\text {max }} / \mathrm{cm}^{-1} 3306(\mathrm{C} \equiv \mathrm{C}-\mathrm{H})$, $2120(\mathrm{C}=\mathrm{C})$ and $1727(\mathrm{CO}) ; \delta_{\mathrm{H}} 7.36-7.28(10 \mathrm{H}, \mathrm{m}$, aromatic H$)$, 5.63-5.57 ( $1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$ ), 4.83-4.77 ( 2 H , m, benzylic H), 4.62$4.57(2 \mathrm{H}, \mathrm{m}$, benzylic H$), 4.36\left(\frac{41}{100} \times 1 \mathrm{H}\right.$, dd, $J 6.4$ and $\left.1.5,9-\mathrm{H}\right)$, $4.33\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.1.5,9-\mathrm{H}\right), 4.31\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $2.4,10-\mathrm{H}), 4.28\left(\frac{59}{100} \times 1 \mathrm{H}\right.$, dd, $J 6.4$ and $\left.2.4,10-\mathrm{H}\right), 3.71$ $(2 \mathrm{H}, \mathrm{dt}, J 7.8$ and $4.9,1-\mathrm{H}), 3.65-3.49(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H})$,
$3.40\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.40\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.36\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}\right.$, $\mathrm{Me}), 3.36\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 2.53\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{d}, J 2.4,12-\mathrm{H}\right), 2.51$ $\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{d}, J 2.4,12-\mathrm{H}\right), 2.12-1.49(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 1.24$ $\left(\frac{59}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right), 1.21\left(\frac{41}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right), 0.90\left(\frac{59}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right)$, $0.90\left(\frac{41}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right), 0.06\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 0.06\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}\right.$, $\mathrm{Me}), 0.06\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right)$ and $0.05\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right) ; m / z 664$ ( $\mathrm{M}^{+}, 1.1 \%$ ), 607 (5.5), 519 (13), 461 (11), 145 (35), 91 (100), 89 (57) and 57 (60).

## ( $3 S, 4 S, 6 R, 9 S, 10 S$ )- and ( $3 S, 4 S, 6 S, 9 S, 10 S$ )-9,10-Bis(benzyl-oxy)-3,4-dimethoxy-6-(pivaloyloxy)dodeca-7,11-diyn-1-ol 30

To a solution of $\mathbf{2 9}(308 \mathrm{mg}, 0.46 \mathrm{mmol})$ in THF ( $2.0 \mathrm{~cm}^{3}$ ) was added $75 \%$ aq. $\mathrm{AcOH}\left(8 \mathrm{~cm}^{3}\right)$ at rt. The reaction mixture was stirred for 24 h at rt , and quenched by addition of $10 \%$ aq. NaOH and extracted with AcOEt. The extract was washed with water and brine, dried and concentrated to dryness. The residue was chromatographed with hexane- $\operatorname{AcOEt}$ (2:1) to give $\mathbf{3 0}$ (229 $\mathrm{mg}, 90 \%$ ) as a colorless oil (Found: $\mathrm{M}^{+}$, 550.2933. $\mathrm{C}_{33} \mathrm{H}_{42} \mathrm{O}_{7}$ requires $M, 550.2931)$; $v_{\text {max }} / \mathrm{cm}^{-1} 3470(\mathrm{OH}), 3306(\mathrm{C} \equiv \mathrm{C}-\mathrm{H})$, $2120(\mathrm{C} \equiv \mathrm{C})$ and $1728(\mathrm{CO}) ; \delta_{\mathrm{H}} 7.36-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$)$, 5.63-5.58 ( $1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$ ), 4.82-4.78 ( $2 \mathrm{H}, \mathrm{m}$, benzylic H), 4.62$4.57(2 \mathrm{H}, \mathrm{m}$, benzylic H$), 4.36\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.1.5,9-\mathrm{H}\right)$, $4.33\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.1.5,9-\mathrm{H}\right), 4.31\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $2.0,10-\mathrm{H}), 4.29\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.2.0,10-\mathrm{H}\right), 3.79-$ $3.71(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 3.71-3.49(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.43$ $\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.42\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.38\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right)$, $3.37\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 2.54\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{d}, J 2.0,12-\mathrm{H}\right), 2.53\left(\frac{41}{100} \times\right.$ $1 \mathrm{H}, \mathrm{d}, J 2.0,12-\mathrm{H}), 2.47\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}\right), 2.46\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{br}\right.$ $\mathrm{s}, \mathrm{OH}), 2.14-1.63(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 1.24\left(\frac{41}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{\mathrm{t}} \mathrm{Bu}\right)$ and $1.21\left(\frac{59}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right) ; m / z 550\left(\mathrm{M}^{+}, 0.7 \%\right), 461(51), 405$ (48), 91 (83) and 57 (100).

## ( $3 S, 4 S, 6 R, 9 S, 10 S$ )- and ( $3 S, 4 S, 6 S, 9 S, 10 S$ )-9,10-Bis(benzyl-oxy)-12-iodo-3,4-dimethoxy-6-(pivaloyloxy)dodeca-7,11-diyn-1-ol 31

To a solution of $\mathbf{3 0}\left(9.70 \mathrm{mg}, 17.6 \times 10^{-3} \mathrm{mmol}\right)$ in DMF $(0.3$ $\mathrm{cm}^{3}$ ) was successively added NIS ( $12.0 \mathrm{mg}, 52.8 \times 10^{-3} \mathrm{mmol}$ ) and $\mathrm{AgNO}_{3}\left(3.00 \mathrm{mg}, 17.6 \times 10^{-3} \mathrm{mmol}\right)$ at rt . The reaction mixture was stirred at rt for 3.5 h and cooled down to $0^{\circ} \mathrm{C}$. Cold water was added to the reaction mixture, which was then extracted with ether several times. The combined layers were washed with brine, dried, and concentrated to dryness. The residue was chromatographed with hexane-AcOEt $(2: 1)$ to give $31(10.5 \mathrm{mg}, 88 \%)$ as a pale yellow oil (Found: $\mathrm{M}^{+}+23$, 699.1754. $\mathrm{C}_{33} \mathrm{H}_{41} \mathrm{IO}_{7} \mathrm{Na}$ requires $\left.M+23,699.1795\right) ; v_{\text {max }} / \mathrm{cm}^{-1}$ $3470(\mathrm{OH}), 2200(\mathrm{C} \equiv \mathrm{C})$ and $1728(\mathrm{CO}) ; \delta_{\mathrm{H}} 7.36-7.27(10 \mathrm{H}, \mathrm{m}$, aromatic H), $5.62-5.59(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 4.81-4.76(2 \mathrm{H}, \mathrm{m}$, benzylic H), 4.61-4.56 ( $2 \mathrm{H}, \mathrm{m}$, benzylic H), $4.42\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{d}\right.$, $J 6.4,10-\mathrm{H}), 4.40\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{d}, J 6.4,10-\mathrm{H}\right), 4.32\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{dd}\right.$, $J 6.4$ and $1.5,9-\mathrm{H}), 4.30\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.1.5,9-\mathrm{H}\right), 3.77-$ $3.73(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 3.68-3.44(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.45$ $\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.43\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.39\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right)$, $3.39\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 2.55-2.40(1 \mathrm{H}, \mathrm{br}$ s, OH), 2.16-1.68 (4H, $\mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}), 1.25\left(\frac{41}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right)$ and $1.22\left(\frac{59}{100} \times 9 \mathrm{H}\right.$, s , $\left.{ }^{t} \mathrm{Bu}\right)$; FABMS $m / z 699\left(\mathrm{M}^{+}+23,2.5 \%\right), 101$ (22), 91 (100) and 57 (44).

## ( $3 S, 4 S, 6 R, 9 S, 10 S$ )- and ( $3 S, 4 S, 6 S, 9 S, 10 S$ )-9,10-Bis(benzyl-oxy)-12-iodo-3,4-dimethoxy-6-(pivaloyloxy)dodeca-7,11-diynal 32

PCC ( $6.50 \mathrm{mg}, 30.1 \times 10^{-3} \mathrm{mmol}$ ) and MS $4 \AA(50.0 \mathrm{mg})$ was added in one portion to a stirred solution of $31(5.10 \mathrm{mg}$, $\left.7.53 \times 10^{-3} \mathrm{mmol}\right)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1.0 \mathrm{~cm}^{3}\right)$ at rt . After being stirred for 3 h , the reaction mixture was filtered through Florisil and washed with ether several times. The filtrate was concentrated to leave the residue which was chromatographed with hexaneAcOEt (4:1) to give $31(4.70 \mathrm{mg}, 93 \%)$ as a pale yellow oil (Found: $\mathrm{M}^{+}+23,697.1644 . \mathrm{C}_{33} \mathrm{H}_{39} \mathrm{IO}_{7} \mathrm{Na}$ requires $M+23$,
697.1638); $v_{\max } / \mathrm{cm}^{-1} 2200(\mathrm{C}=\mathrm{C})$ and $1728(\mathrm{CO}) ; \delta_{\mathrm{H}} 9.81(1 \mathrm{H}$, dd, $J 3.9$ and 1.0, CHO), $7.36-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H), $5.61-$ $5.58(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 4.81-4.77(2 \mathrm{H}, \mathrm{m}$, benzylic H), 4.61-4.56 $(2 \mathrm{H}, \mathrm{m}$, benzylic H$), 4.42\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{d}, J 6.4,10-\mathrm{H}\right), 4.40$ $\left(\frac{41}{100} \times 1 \mathrm{H}, \mathrm{d}, J 6.4,10-\mathrm{H}\right), 4.32\left(\frac{59}{100} \times 1 \mathrm{H}, \mathrm{dd}, J 6.4\right.$ and $\left.2.0,9-\mathrm{H}\right)$, $4.30\left(\frac{41}{100} \times 1 \mathrm{H}\right.$, dd, $J 6.4$ and $\left.2.0,9-\mathrm{H}\right), 3.96-3.44(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.40\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.40\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.37$ $\left(\frac{59}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 3.35\left(\frac{41}{100} \times 3 \mathrm{H}, \mathrm{s}, \mathrm{Me}\right), 2.69-2.51(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, $2.17-1.85(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 1.25\left(\frac{41}{100} \times 9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right)$ and $1.23\left(\frac{59}{100} \times\right.$ $\left.9 \mathrm{H}, \mathrm{s},{ }^{\dagger} \mathrm{Bu}\right) ;$ FABMS $m / z 697\left(\mathrm{M}^{+}+23,1.3 \%\right), 136$ (12), 91 (100) and 57 (58).

## $\mathrm{CrCl}_{2}$-mediated coupling reaction of $\mathbf{3 2}$

A solution of $\mathbf{3 2}(77.5 \mathrm{mg}, 0.12 \mathrm{mmol})$ in THF $\left(15 \mathrm{~cm}^{3}\right)$ was added over a period of 15 min to a suspension of $\mathrm{CrCl}_{2}(141$ $\mathrm{mg}, 1.15 \mathrm{mmol})$ and $\mathrm{NiCl}_{2}\left(7.40 \mathrm{mg}, 57 \times 10^{-3} \mathrm{mmol}\right)$ in THF $\left(100 \mathrm{~cm}^{3}\right)$ at rt . The mixture was stirred for 24 h at rt , diluted with ether which was washed with water and brine, dried and concentrated to dryness. The residue was chromatographed with hexane-AcOEt (3:1) to give $(3 S, 4 S, 9 S, 10 S)$ - 9,10 -bis-(benzyloxy)-3,4-dimethoxy-6-(pivaloyloxy)cylcododeca-7,11-diyn-1-ol (33) ( $34.1 \mathrm{mg}, 54 \%$ ) as a mixture of stereoisomers. Compound 33 was a colorless oil (Found: $\mathrm{M}^{+}$, 548.2778. $\mathrm{C}_{33} \mathrm{H}_{40} \mathrm{O}_{7}$ requires $\left.M, 548.2774\right)$; $v_{\text {max }} / \mathrm{cm}^{-1} 3610(\mathrm{OH}), 3420$ $(\mathrm{OH})$ and $1728(\mathrm{CO}) ; \delta_{\mathrm{H}} 7.37-7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$)$, $5.53-5.63(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 4.79-4.74(2 \mathrm{H}, \mathrm{m}$, benzylic H), 4.70$4.66(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 4.63-4.55(2 \mathrm{H}, \mathrm{m}$, benzylic H), 4.40-4.34(2H, $\mathrm{m}, 9-\mathrm{H}$ and $10-\mathrm{H}$ ), $3.92-3.58(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.45-3.41$ $(6 \mathrm{H}, \mathrm{m}, \mathrm{Me}), 2.50-2.35(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 2.28-1.89(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H})$ and $1.24\left(9 \mathrm{H}, \mathrm{s},{ }^{t} \mathrm{Bu}\right) ; m / z 548\left(\mathrm{M}^{+}, 0.4 \%\right), 153(3.4)$, 91 (100), 77 (6.7) and 57 (20).

## Reaction of 33 with DIBAL-H

A solution of DIBAL-H in hexane $\left(0.95 \mathrm{~mol} \mathrm{dm}^{-3} ; 0.04 \mathrm{~cm}^{3}\right.$, $\left.41.6 \times 10^{-3} \mathrm{mmol}\right)$ was added to a solution of $33(5.70 \mathrm{mg}$, $\left.10.3 \times 10^{-3} \mathrm{mmol}\right)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1.0 \mathrm{~cm}^{3}\right)$ at $-78{ }^{\circ} \mathrm{C}$. The mixture was stirred for 5 min , quenched by addition of saturated aq. $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered through Celite. The filtrate was concentrated to leave the residue, which was chromatographed with hexane-AcOEt ( $1: 3$ ) to give ( $3 S, 4 S, 9 S, 10 S$ )-9, 10 -bis(benzyl-oxy)-3,4-dimethoxycyclododeca-7,11-diyne-1,6-diol (34) (4.50 $\mathrm{mg}, 94 \%$ ) as a mixture of stereoisomers. Compound 34 was a colorless oil (Found: $\mathrm{M}^{+}, 464.2201 . \mathrm{C}_{28} \mathrm{H}_{32} \mathrm{O}_{6}$ requires $M$, 464.2199); $v_{\max } / \mathrm{cm}^{-1} 3610(\mathrm{OH})$ and $3431(\mathrm{OH}) ; \delta_{\mathrm{H}} 7.36-$ $7.26(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 4.80-4.75(2 \mathrm{H}, \mathrm{m}$, benzylic H$)$, 4.69-4.65 ( $1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ and $6-\mathrm{H}), 4.63-4.58(2 \mathrm{H}, \mathrm{m}$, benzylic H), $4.44-4.35(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ and $10-\mathrm{H}), 4.08-3.66(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.48-3.43(6 \mathrm{H}, \mathrm{m}, \mathrm{Me})$ and 2.24-1.95 (4H, m, 2-H and $5-\mathrm{H}) ; m / z 464\left(\mathrm{M}^{+}, 0.7 \%\right), 446$ (0.9), 105 (7.9), 91 (100) and 77 (9.7).

## (3S,4S,9S,10S)-9,10-Bis(benzyloxy)-3,4-dimethoxycyclododeca-7,11-diyne-1,6-dione (+)-35

According to the procedure described for conversion of 31 to 32, $34\left(1.50 \mathrm{mg}, 3.23 \times 10^{-3} \mathrm{mmol}\right)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1.0 \mathrm{~cm}^{3}\right)$ was oxidized with PCC ( $11.0 \mathrm{mg}, 51.0 \times 10^{-3} \mathrm{mmol}$ ) in the presence of MS $4 \AA(50.0 \mathrm{mg})$. Work-up gave the residue which was chromatographed with hexane-AcOEt (3:1) to give $(+)-35$ $(1.20 \mathrm{mg}, 81 \%)$ as a colorless solid $\mathrm{mp} 94-95^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: $\mathrm{M}^{+}, 460.1882 . \mathrm{C}_{28} \mathrm{H}_{28} \mathrm{O}_{6}$ requires $M$, 460.1886); $[a]_{\mathrm{D}}^{25}+50.8\left(c 0.20, \mathrm{CHCl}_{3}\right) ; v_{\max } / \mathrm{cm}^{-1} 2220(\mathrm{C} \equiv \mathrm{C})$ and $1672(\mathrm{CO}) ; \delta_{\mathrm{H}} 7.37-7.29(10 \mathrm{H}, \mathrm{m}$, aromatic H$), 4.80,4.60$ $(4 \mathrm{H}, \mathrm{AB}-\mathrm{q}, J 11.7$, benzylic H), $4.55(4 \mathrm{H}, \mathrm{s}, 9-\mathrm{H}$ and $10-\mathrm{H}), 4.10$ $(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}$ and $4-\mathrm{H}), 3.37(6 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.03(2 \mathrm{H}, \mathrm{dd}, J 15.6$ and $9.3,2-\mathrm{H}$ and $5-\mathrm{H})$ and $2.86(2 \mathrm{H}, \mathrm{dd}, J 15.6$ and $3.9,2-\mathrm{H}$ and $5-\mathrm{H}) ; \delta_{\mathrm{C}} 184.2,136.1,128.6,128.4,128.0,89.5,85.4,78.1,72.1$, $71.4,58.1$ and $45.4 ; m / z 460\left(\mathrm{M}^{+}, 9.3 \%\right), 369$ (14), 337 (14), 139 (18) and 91 (100).

## References

1 N. Ishida, K. Miyazaki, K. Kumagai and M. Rikimaru, J. Antibiot., 1965, 18, 68.
2 K. Edo, M. Mizugaki, Y. Koide, H. Seto, K. Furihata, N. Otake and N. Ishida, Tetrahedron Lett., 1985, 26, 331.

3 K. S. Lam, G. A. Hesler, D. R. Gustavson, A. R. Crosswell, J. M. Veitch, S. Forenza and K. Tomita, J. Antibiot., 1991, 44, 472.

4 J. Hu, Y.-C. Xue, M.-Y. Xie, Y. Zhang, T. Otani, Y. Minami, Y. Yamada and T. Marunaka, J. Antibiot., 1988, 41, 1575.

5 M. Hanada, H. Ohkuma, T. Yonemoto, K. Tomita, M. Ohbayashi, H. Kamei, T. Miyaki, M. Konishi, H. Kawaguchi and S. Forenza, J. Antibiot., 1991, 44, 403.

6 T. Ando, M. Ishii, T. Kajiura, Y. Kameyama, K. Miwa and Y. Sugiura, Tetrahedron Lett., 1998, 39, 6495.
7 (a) P. A. Wender, M. Harmata, D. Jeffrey, C. Mukai and J. Suffert, Tetrahedron Lett., 1988, 29, 909; (b) P. A. Wender, J. A. McKinney and C. Mukai, J. Am. Chem. Soc., 1990, 112, 5369; (c) P. Magnus and M. Davis, J. Chem. Soc., Chem. Commun., 1991, 1522; (d) T. Takahashi, H. Tanaka, Y. Hirai, T. Doi, H. Yamada, T. Shiraki and Y. Sugiura, Angew. Chem., Int. Ed. Engl., 1993, 32, 1657; (e) A. G. Myers, M. Hammond, Y. Wu, J.-N. Xiang, P. M. Harrington and E. Y. Kuo, J. Am. Chem. Soc., 1996, 118, 10006; ( $f$ ) P. Magnus, R. Carter, M. Davis, J. Elliott and T. Pitterna, Tetrahedron Lett., 1996, 37, 6283; (g) S. Caddick and V. M. Delisser, Tetrahedron Lett., 1997, 38, 2355.
8 (a) M. Hirama, K. Fujiwara, K. Shigematu and Y. Fukazawa, J. Am. Chem. Soc., 1989, 111, 4120; (b) M. Hirama, T. Gomibuchi and K. Fujiwara, J. Am. Chem. Soc., 1991, 113, 9851; (c) K. Nakatani, K. Arai and S. Terashima, J. Chem. Soc., Chem. Commun., 1992, 289; (d) J. Suffert, E. Abraham, S. Raeppel and R. Brückner, Liebigs Ann. Chem., 1996, 334; (e) K. Brickmann, F. Hambloch, J. Suffert and R. Brückner, Liebigs Ann. Chem., 1996, 357; (f) M. Eckhardt and R. Brückner, Liebigs Ann. Chem., 1996, 473 and references cited therein.
9 During this investigation, Myers and Goldberg have recently reported a novel transformation reaction of the twelve-membered tetrayne derivative into the corresponding bicyclo[7.3.0]dodecadienediyne: A. G. Myers and S. D. Goldberg, Tetrahedron Lett., 1998, 39, 9633.
10 For recent reviews on medium-sized cyclic diynes, see (a) R. Gleiter and R. Merger, in Cyclic Alkynes, Preparation and Properties in Modern Acetylene Chemistry, ed. P. J. Stang and F. Diederich, VCH, Weinheim, 1995, p. 285; (b) C. Boss and R. Keese, Tetrahedron, 1997, 53, 3111 and references cited therein.
11 J. Kuszmann, Carbohydr. Res., 1979, 71, 123
12 M. Yamaguchi and I. Hirao, Tetrahedron Lett., 1983, 24, 391.
13 B. B. Lohray, Y. Jayamma and M. Chatterjee, J. Org. Chem., 1995, 60, 5958.
14 J. E. Baldwin, J. Chem. Soc., Chem. Commun., 1976, 734.
15 H. Nemoto, S. Takamatsu and Y. Yamamoto, J. Org. Chem., 1991, 56, 1321.
16 E. J. Corey and P. L. Fuchs, Tetrahedron Lett., 1972, 13, 3769.
17 Y. L. Merrer, A. Dureault, C. Greck, D. M. Languin, C. Gravier and J.-C. Depezay, Heterocycles, 1987, 25, 541.

18 Numbering for NCS chromophore (1) was employed for convenience.
19 The structure of $\mathbf{2 3}$ was elucidated by spectral evidence. The IR spectrum showed a diagnostical absorption band at $1952 \mathrm{~cm}^{-1}$, a typical allenic band, along with a carbonyl absorption at 1675 $\mathrm{cm}^{-1}$. In the ${ }^{1} \mathrm{H}$ NMR spectrum, two characteristically lower field-shifted peaks appeared at $\delta 5.97(\mathrm{dd}, J=6.3$ and 3.3 Hz$)$ and 5.91 (dd, $J=6.3$ and 4.6 Hz ) attributable to $\mathrm{H}-8^{18}$ and $\mathrm{H}-6$, respectively. The decoupling experiments disclosed that H-5 coupled not only with H-8 ( $J=3.3 \mathrm{~Hz}$ ), but also with H-6 $(J=4.6 \mathrm{~Hz})$ strongly supporting the structure. Furthermore, a diagnostic peak due to the allenic carbon at $\delta 213.6$ was observed in the ${ }^{13} \mathrm{C}$ NMR spectrum.
20 A similar isomerization from the propargyl (prop-2-ynyl) ketone moiety to the corresponding allenyl ketone functionality was reported: (a) G. Majetich, Y. Zhang and G. Dreyer, Tetrahedron Lett., 1993, 34, 449; (b) A. S. K. Hashmi, J. W. Bats, J.-H. Choi and L. Schwarz, Tetrahedron Lett., 1998, 39, 7491 and references cited therein.
21 A. C. Cope and A. S. Mehta, J. Am. Chem. Soc., 1964, 86, 5626.
22 (a) A. G. Myers, P. M. Harrington and E. Y. Kuo, J. Am. Chem. Soc., 1991, 113, 694; (b) K. Iida and M. Hirama, J. Am. Chem. Soc., 1994, 116, 10310; (c) I. Sato, K. Toyama, T. Kikuchi and M. Hirama, Synlett, 1998, 1308.

23 (a) K. C. Nicolaou, A. Liu, Z. Zeng and S. McComb, J. Am. Chem. Soc., 1992, 114, 9279; (b) C. Meert, J. Wang and P. J. De Clercq, Tetrahedron Lett., 1997, 38, 2179; (c) M. F. Semmelhack, Y. Gu and D. M. Ho, Tetrahedron Lett., 1997, 38, 5583.

24 T. Nishizawa, S. Shibuya, S. Hosokawa and M. Isobe, Synlett, 1994 485.

25 Several attempts were made to develop a more convenient procedure involving fewer steps for obtaining 34. For example, the dialdehyde compound derived from 26 (e.g., compound 6 in Scheme 1) was reacted with dicerium acetylide or diiodoacetylene derivatives, prepared from 13, to afford an intractable mixture.

26 Several preliminary attempts on conversion of $\mathbf{3 5}$ into the corresponding bicyclo[7.3.0]dodecadiyne derivatives were made under typical aldol conditions by using LDA or LHMDS. However, no ring closed compounds have so far been isolated from the reaction mixture.

Paper a909089d

